CCLXXVIII.—Preparation of 3:5-Dihalogenophenols.

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The whole series of 3:5-dihalogenophenols required in the course of a general investigation of symmetrical substitution in benzene has been prepared from 1:3:5-trinitrobenzene. Some of them and their derivatives were already known (in most cases the m. p.'s recorded in the literature have been confirmed); the new compounds are described below.

1:3:5-Trinitrobenzene was prepared by Clark and Hartmann's method ("Organic Syntheses," Vol. II, pp. 93—97), by oxidising 2:4:6-trinitrotoluene (360 g.) below 45°; a yield of 210 g. was obtained.

5-Nitro-3-aminoanisole was obtained in 85% yield by boiling a mixture of 3:5-dinitroanisole (30 g.), sodium trisulphide (18 g.), and alcohol (150 c.c.) for 4 hours, adding water, and extracting the precipitate with boiling dilute hydrochloric acid. It crystallised from water in orange needles, m. p. 120° (Blanksma, *Rec. trav. chim.*, 1905, 24, 40, gives m. p. 118°; Meister, Lucius, & Brüning, D.R.-P. 22062, give 120°).

The 3-halogeno-5-nitroanisoles were obtained from the preceding compound by the Sandmeyer method, and all crystallised from alcohol in colourless needles. 3-Iodo-5-nitroanisole has m. p. 84° (Found: I, 45·2. $C_7H_6O_3NI$ requires I, 45·5%).

The 3-halogeno-5-nitrophenols were prepared by heating the 3-halogeno-5-nitroanisoles (10 g.) with concentrated sulphuric acid (30 c.c.) for 2 hours, pouring the mixture on to ice, and crystallising the 3-halogeno-5-nitrophenol from hot dilute mineral acid (yield about 70%). The benzoates crystallise from alcohol, the acetates from water or dilute alcohol, and the 2:4:6-tribromo-derivatives from dilute acetic acid or alcohol, all in colourless needles. 5-nitrophenyl benzoate has m. p. 78° (Found : Cl, 12.6. $C_{13}H_8O_4NCl$ requires Cl, 12.8%), and the acetate has m. p. 84° (Found: Cl, 16.3. C₈H₆O₄NCl requires Cl, 16.5%). 3-Bromo-5-nitrophenyl acetate has m. p. 99° (Found : Br, 30.6. $C_8H_6O_4NBr$ requires Br, 30.8%). 3-Iodo-5-nitrophenol crystallises from carbon tetrachloride or water in needles, m. p. 136° (Found: I, 47.7. C_eH₄O₃NI requires I, 47.9%), the pale yellow colour of which remains unchanged after repeated crystallisations. The benzoate has m. p. 100.5° (Found: I, 34·1. C₁₃H₈O₄NI requires I, 34·4%), and the acetate, m. p. 110° (Found: I, $41 \cdot 1$. $C_8 H_6 O_4 NI$ requires I, $41 \cdot 3\%$). $2 : 4 : 6 \cdot Tri$ bromo-3-iodo-5-nitrophenol forms colourless plates, m. p. 176°

(0·1337 g. gave 0·2113 g. of mixed silver halides. Required, 0·2127 g.).

3-Halogeno-5-aminoanisoles.—The chloro- and bromo-compounds were readily prepared by reducing the corresponding nitro-compounds with tin and hydrochloric acid. The following process was more serviceable in the case of 3-iodo-5-aminoanisole: 3-Iodo-5-nitroanisole (12 g.), suspended in a mixture of water (200 c.c.) and concentrated hydrochloric acid (15 c.c.), was heated on the water-bath during the gradual addition of iron filings (24 g.). After 30 minutes, the mixture was rendered alkaline, and the 3-iodo-5-aminoanisole distilled in steam. All three compounds crystallise from light petroleum in colourless needles. 3-Chloro-5-aminoanisole has m. p. 33° (Found: Cl, 22·5. C₇H₈ONCl requires Cl, 22·5%), 3-bromo-5-aminoanisole, m. p. 52° (Found: Br, 39·4. C₇H₈ONBr requires 39·6%), and 3-iodo-5-aminoanisole, m. p. 86·5° (Found: I, 50·7. C₇H₈ONI requires I, 51·0%).

The 3:5-dihalogenoanisoles were prepared from the 3-halogeno-5-aminoanisoles by the Sandmeyer method. 3-Chloro-5-iodoanisole is a liquid, b. p. 267—268°, which solidifies at 0° (0·1882 g. gave 0·2672 g. of silver halides. Required, 0·2652 g.). The others all crystallise from light petroleum in colourless aggregates. 3:5-Diiodoanisole has m. p. 85° (Found: I, 70·5. $C_7H_6OI_2$ requires I, 70·5%), 3-chloro-5-bromoanisole, m. p. 33° (0·1026 g. gave 0·1535 g. of silver halides. Required, 0·1534 g.), and 3-bromo-5-iodoanisole, m. p. 54° (0·1219 g. gave 0·1660 g. of silver halides. Required, 0·1646 g.).

3:5-Dihalogenophenols.—It was found more convenient to hydrolyse the 3-halogeno-5-aminoanisoles with concentrated sulphuric acid as described above and to apply the Sandmeyer process to the diluted diazotised solution than to replace the aminogroup by halogen and hydrolyse the 3:5-dihalogenoanisole.

The 3:5-dihalogenophenols are moderately easily volatile in steam and crystallise from light petroleum in colourless needles. The benzoates, acetates, and tribromo-derivatives all crystallise from alcohol or dilute alcohol in colourless needles. 3:5-Dichlorophenyl benzoate has m. p. 55° (Found: Cl, 26·5. $C_{13}H_8O_2Cl_2$ requires Cl, 26·6%), and the acetate, m. p. 38° (Found: Cl, 34·3. $C_8H_6O_2Cl_2$ requires Cl, 34·6%).

3:5-Dibromophenol has m. p. 81° (Blau, Monatsh., 1886,~7,~630, gives m. p. $76\cdot5^\circ$; Blanksma, Rec. trav. chim., 1892,~12,~254, gives 81°). The benzoate has m. p. 77° (Found: Br, $44\cdot8.$ $C_{13}H_8O_2Br_2$ requires Br, $44\cdot9\%$), and the acetate, m. p. 53° (Found: Br, $54\cdot5$. $C_8H_6O_2Br_2$ requires Br, $54\cdot4\%$). $3:5\text{-}Di\text{-}iodophenylbenzoate}$ has m. p. 93° (Found: I, $56\cdot2$. $C_{13}H_8O_2I_2$ requires I,

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 $56\cdot 4\%).$ 2 : 4 : 6-Tribromo-3 : 5-di-iodophenol has m. p. 226—228° (0·1166 g.; 0·2059 g.; 0·2068 g.*).

3-Chloro-5-bromophenol has m. p. 70° (0.0728 g.; 0.1157 g.; 0.1162 g.), the benzoate m. p. 62° (0.1016 g.; 0.1072 g.; 0.1080 g.), and the acetate m. p. 45° (0.0716 g.; 0.0911 g.; 0.0950 g.). 3-Chloro-2:4:5:6-tetrabromophenol has m. p. 205° (0.0657 g.; 0.1312 g.; 0.1323 g.). 3-Chloro-5-iodophenol has m. p. 60° (0.1273 g.; 0.2385 g.; 0.2393 g.), the benzoate m. p. 54° (0.1793 g.; 0.2387 g.; 0.2393 g.), and the acetate m. p. 47° (0.1483 g.; 0.2395 g.; 0.2393 g.). 3-Chloro-2:4:6-tribromo-5-iodophenol has m. p. 195° (0.1229 g.; 0.2350 g.; 0.2356 g.).

3-Bromo-5-iodophenol has m. p. $82\cdot5^\circ$ (0·1013 g.; 0·1430 g.; 0·1432 g.), the benzoate m. p. 76° (0·1066 g.; 0·1108 g.; 0·1119 g.), and the acetate m. p. 46° (0·0723 g.; 0·0884 g.; 0·0896 g.). 2:3:4:6-Tetrabromo-5-iodophenol has m. p. 220— 221° (0·1072 g.; 0·1962 g.; 0·1974 g.).

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